



NSW Education Standards Authority

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Centre Number

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Student Number

2023 HIGHER SCHOOL CERTIFICATE EXAMINATION

Physics

General Instructions

- Reading time – 5 minutes
- Working time – 3 hours
- Write using black pen
- Draw diagrams using pencil
- Calculators approved by NESA may be used
- A data sheet, formulae sheet and Periodic Table are provided at the back of this paper
- Write your Centre Number and Student Number at the top of this page

Total marks: 100

Section I – 20 marks (pages 2–13)

- Attempt Questions 1–20
- Allow about 35 minutes for this section

Section II – 80 marks (pages 17–39)

- Attempt Questions 21–34
- Allow about 2 hours and 25 minutes for this section

Section I

20 marks

Attempt Questions 1–20

Allow about 35 minutes for this section

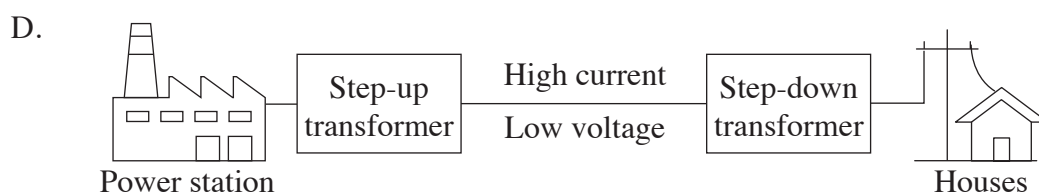
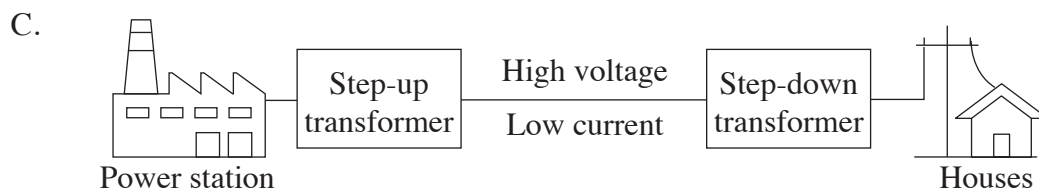
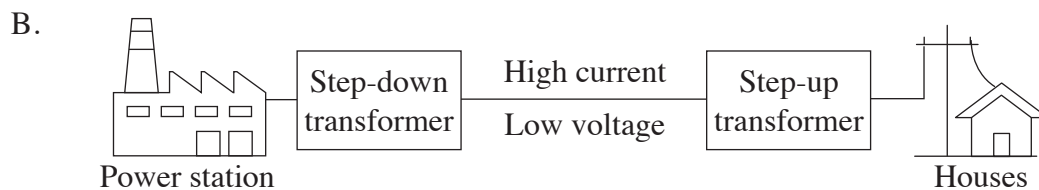
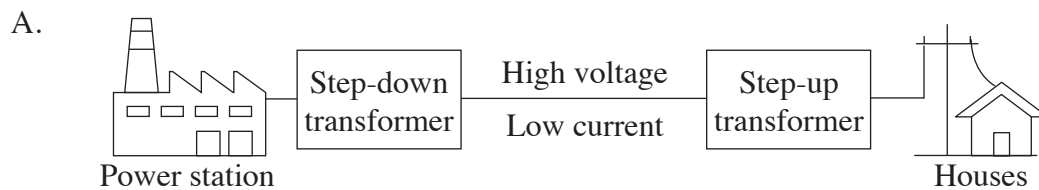
Use the multiple-choice answer sheet for Questions 1–20.

- 1 The gravitational field strength acting on a spacecraft decreases as its altitude increases.

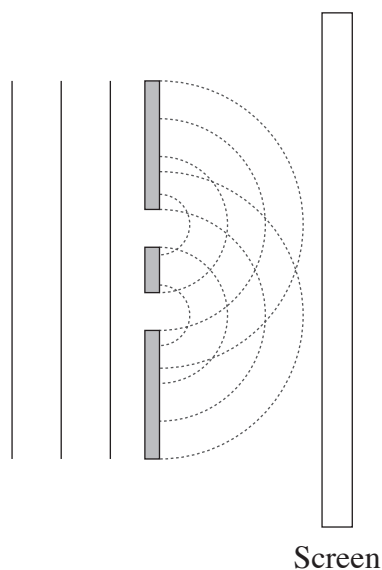
This is due to a change in the

- A. mass of Earth.
- B. mass of the spacecraft.
- C. density of the atmosphere.
- D. distance of the spacecraft from Earth's centre.




- 2 Which diagram best represents the transmission of energy from a power station to people's houses?



3 A diagram representing a double slit experiment using light is shown.



Which of the following best represents the expected pattern on the screen?

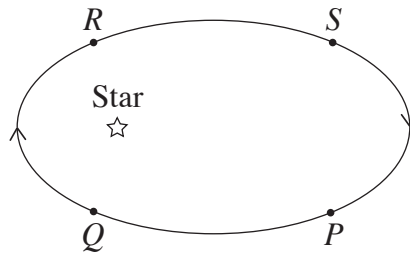
- A. 
- B. 
- C. 
- D. 

4 Caesium-137 has a half-life of 30 years.

What mass of caesium-137 will remain after 90 years, if the initial mass was 120 g?

- A. 4 g
- B. 15 g
- C. 40 g
- D. 60 g

- 5 An exoplanet is in an elliptical orbit, moving in the direction shown. The distances between consecutive positions P , Q , R and S are equal.



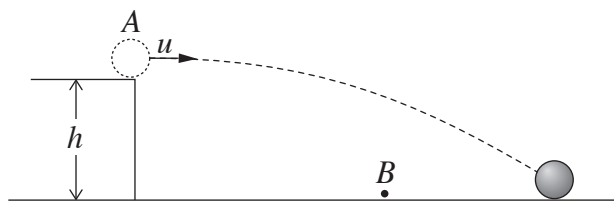
Between which two points is the exoplanet's travel time the greatest?

- A. P and Q
 - B. Q and R
 - C. R and S
 - D. S and P
- 6 An electron would produce an electromagnetic wave when it is
- A. stationary.
 - B. in a stable hydrogen atom.
 - C. moving at a constant velocity.
 - D. moving at a constant speed in a circular path.
- 7 A proton and a neutron travel at the same speed.

Which statement correctly explains the difference between their de Broglie wavelengths?

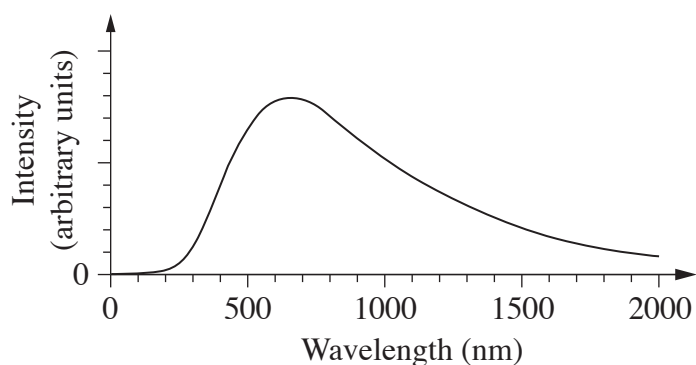
- A. The proton has a longer wavelength because its mass is greater.
- B. The proton has a shorter wavelength because its mass is smaller.
- C. The neutron has a shorter wavelength because its mass is greater.
- D. The neutron has a longer wavelength because its mass is smaller.

- 8 A ball is launched from a platform at position A with velocity u . It lands in the position shown.



The ball could be made to land at position B by increasing the

- A. velocity u .
 - B. launch angle.
 - C. mass of the ball.
 - D. height of the platform.
- 9 The graph shows the relationship between radiation intensity and wavelength for a black body at 4500 K.



Which statement describes the expected difference in the graph for a black body at 4000 K?

- A. Intensity at all wavelengths will be less.
- B. Intensity at all wavelengths will be greater.
- C. The peak intensity will occur at a higher frequency.
- D. The peak intensity will occur at a shorter wavelength.

- 10 Figure I shows a current flowing through a loop of wire that is in a uniform magnetic field.

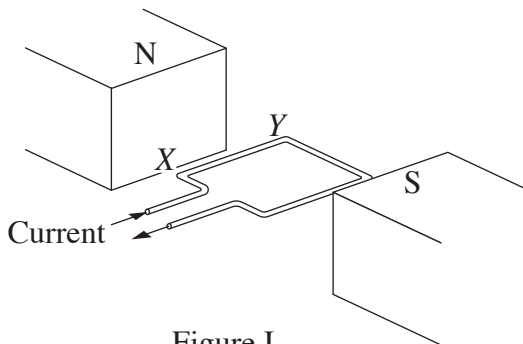


Figure I

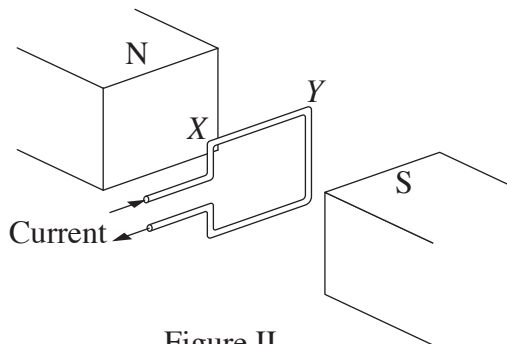


Figure II

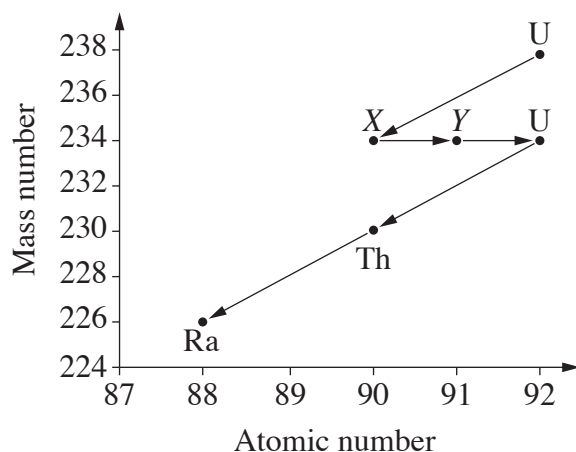
The loop is then rotated to the position shown in Figure II.

The magnitude of the force on the side XY and the magnitude of the torque on the loop in Figure II are compared to those in Figure I.

Which row of the table correctly describes the comparison?

	<i>Force</i>	<i>Torque</i>
A.	$I > II$	$I = II$
B.	$I > II$	$I > II$
C.	$I = II$	$I = II$
D.	$I = II$	$I > II$

- 11 The chart shows part of a nuclear decay series beginning with uranium.



Which option correctly identifies X and Y and the process by which each was produced?

	X	Y
A.	${}_{90}^{234}\text{Th}$ alpha decay	${}_{91}^{234}\text{Pa}$ beta decay
B.	${}_{90}^{234}\text{Th}$ alpha decay	${}_{91}^{234}\text{Pa}$ alpha decay
C.	${}_{91}^{234}\text{Pa}$ beta decay	${}_{90}^{234}\text{Th}$ beta decay
D.	${}_{91}^{234}\text{Pa}$ beta decay	${}_{90}^{234}\text{Th}$ alpha decay

- 12 Figure I shows a positively charged particle accelerating freely from X to Y , between oppositely charged plates. The change in the particle's kinetic energy is W .

The distance between the plates is then doubled as shown in Figure II. The same charge accelerates from rest over the same distance from X to Y .

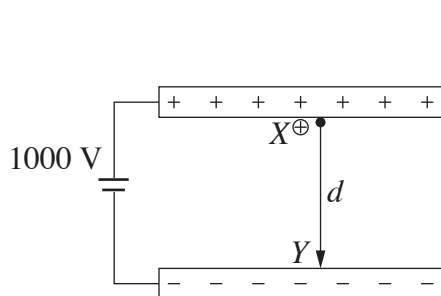


Figure I

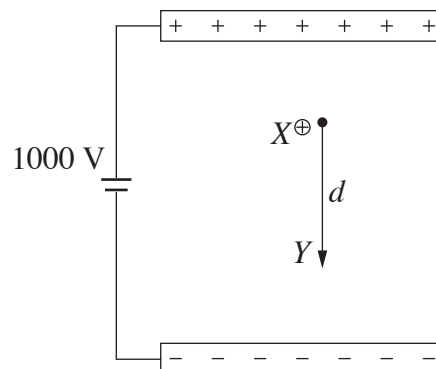


Figure II

What is the change in kinetic energy of the positively charged particle shown in Figure II?

- A. W
 - B. $\frac{W}{2}$
 - C. \sqrt{W}
 - D. $2W$
- 13 Nucleus X has a greater binding energy than nucleus Y .

What can be deduced about X and Y ?

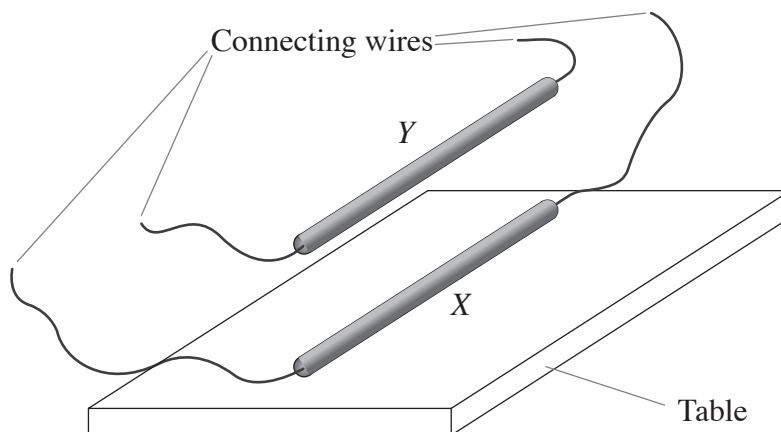
- A. X is more stable than Y .
- B. Y is more stable than X .
- C. X has a greater mass defect than Y .
- D. Y has a greater mass defect than X .

- 14** Planet *X* has a mass 4 times that of Earth and a radius 3 times that of Earth. The escape velocity at the surface of Earth is 11.2 km s^{-1} .

What is the escape velocity at the surface of planet *X*?

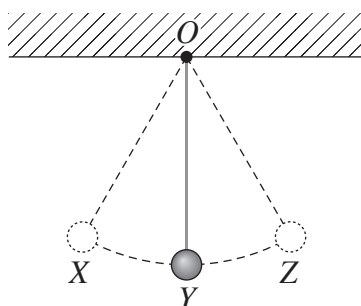
- A. 8.40 km s^{-1}
 - B. 9.70 km s^{-1}
 - C. 12.9 km s^{-1}
 - D. 14.9 km s^{-1}
- 15** What evidence resulting from investigations into the photoelectric effect is consistent with the model of light subsequently proposed by Einstein?
- A. Photoelectrons were only ejected from a metal if the light was less than a specific wavelength.
 - B. Increasing the intensity of light on a metal increased the maximum kinetic energy of the photoelectrons.
 - C. If photons had sufficient energy to eject photoelectrons from a metal, the maximum kinetic energy was independent of the type of metal used.
 - D. The probability of photoelectrons being emitted from a metal was proportional to the duration of exposure to light for any given wavelength used.

- 16 In a thought experiment, two identical parallel aluminium rods, X and Y , are carrying electric currents of equal magnitude. Rod X rests on a table. Rod Y remains stationary, vertically above X , as a result of the magnetic interaction. The masses of the connecting wires are negligible.



Which statement must be correct if rod Y is stationary?

- A. The magnetic force acting on X is upward.
 - B. The currents through X and Y are in the same direction.
 - C. The force the table exerts on X is equal and opposite to the total weight of X and Y .
 - D. The force the table exerts on X is equal and opposite to the force of gravity acting on Y .
- 17 A mass attached to a lightweight, rigid arm hanging from point O , oscillates freely between X and Z .



Which statement best describes the torque acting on the arm as it oscillates?

- A. It is constant in magnitude and direction.
- B. It is zero at Y and a maximum at X and Z .
- C. It is zero at X and Z and a maximum at Y .
- D. It is constant in magnitude but its direction changes.

- 18 The diagrams show the trajectories of two particles with the same mass and charge and which initially have the same velocity u , as shown. The subsequent motion of each particle is determined by its properties and by its interaction with the field in which it is moving.

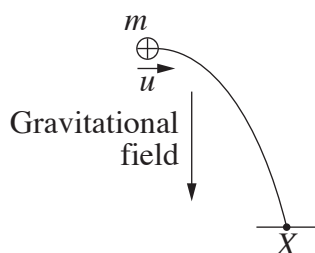


Figure I

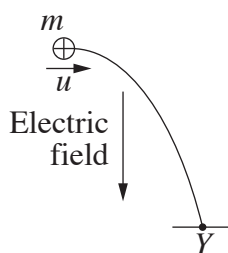


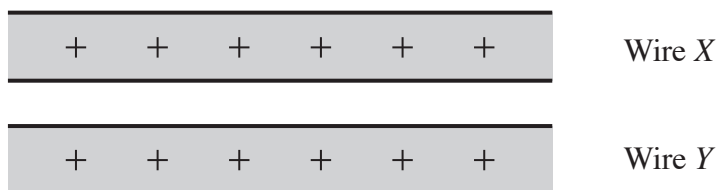
Figure II

X and Y represent the landing points in Figures I and II.

Which row of the table shows the correct paths of the particles if the mass of each is increased by the same amount and they are given the same initial velocity u ?

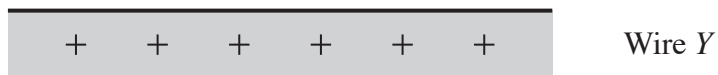
	Gravitational field	Electric field
A.		
B.		
C.		
D.		

- 19 The diagram represents the distribution of positive charges in identical wires when no current is flowing.



Equal currents then flow in each wire, but in opposite directions. These currents are considered conventionally as the flow of positive charge.

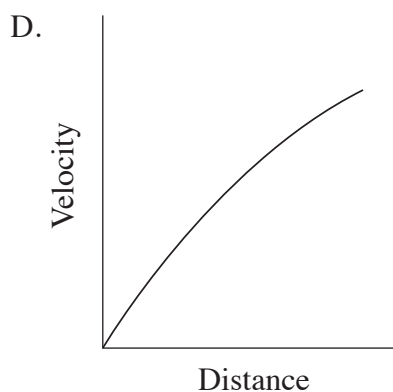
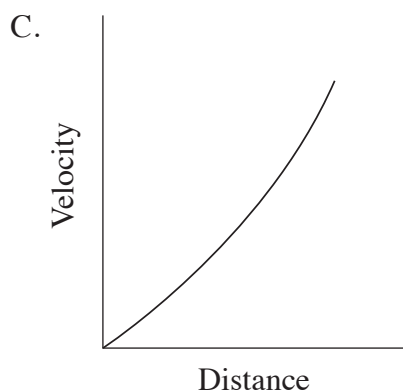
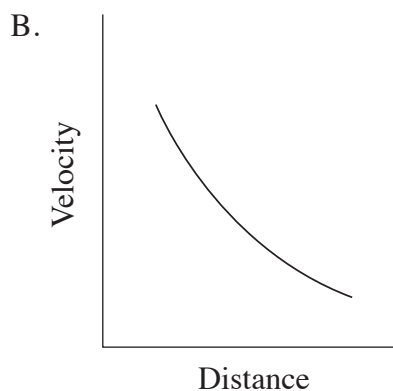
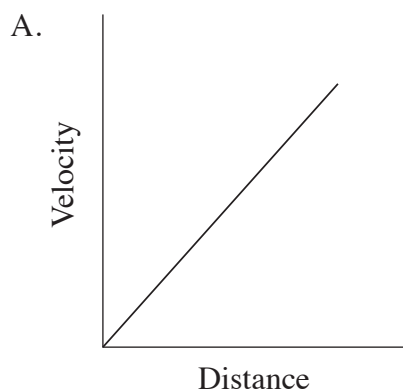
Which diagram represents the charge distribution in the wires, from the frame of reference of a positive charge in wire *Y*?



- 20 In 1995, observational evidence showed that Hubble's description of the expansion of the universe was inaccurate.

It was discovered that the expansion of the universe was accelerating. This discovery was based on observations of light from galaxies whose distances from Earth could be accurately measured, and were significantly more distant than any observed by Hubble.

Which graph relating velocities of galaxies to their distances from Earth is consistent with an accelerating rate of expansion of the universe?



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Centre Number

Physics

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Student Number

Section II Answer Booklet

80 marks

Attempt Questions 21–34

Allow about 2 hours and 25 minutes for this section

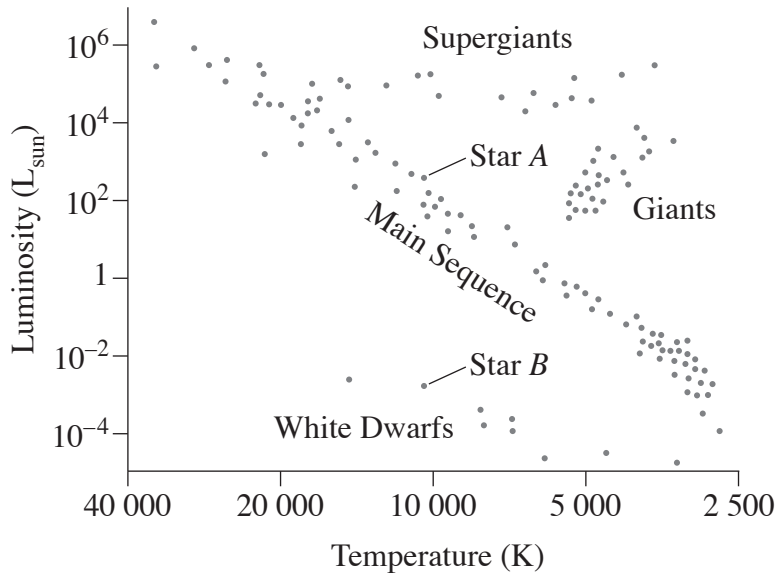
Instructions

- Write your Centre Number and Student Number at the top of this page.
- Answer the questions in the spaces provided. These spaces provide guidance for the expected length of response.
- Show all relevant working in questions involving calculations.
- Extra writing space is provided at the back of this booklet. If you use this space, clearly indicate which question you are answering.

Please turn over

Question 21 (5 marks)

A Hertzsprung–Russell diagram is shown.



- (a) Identify TWO variables that determine the luminosity of a star.

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- (b) Describe differences between stars *A* and *B*.

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Question 22 (3 marks)

A spacecraft passes Earth at a speed of $0.9c$. The spacecraft emits a light pulse every 3.1×10^{-9} s, as measured by the crew on the spacecraft.

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What is the time between the pulses, as measured by an observer on Earth?

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Question 23 (7 marks)

The James Webb Space Telescope (JWST) has a mass of 6.1×10^3 kg and orbits the Sun at a distance of approximately 1.52×10^{11} m.

- (a) The Sun has a mass of 1.99×10^{30} kg. **2**

Calculate the magnitude of gravitational force the Sun exerts on the JWST.

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- (b) The telescope is sensitive to wavelengths from 6.0×10^{-7} m to 2.8×10^{-5} m. **3**

What is the minimum photon energy that it can detect?

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- (c) The JWST observed an exoplanet emitting a peak wavelength of 1.14×10^{-5} m. **2**

Calculate the temperature of the exoplanet.

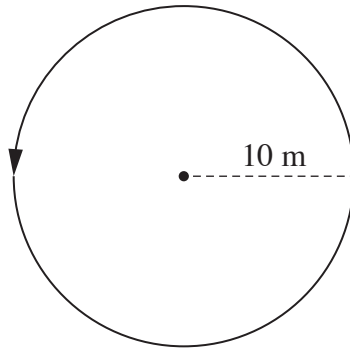
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Question 24 (3 marks)

An electron is travelling at $3.0 \times 10^6 \text{ m s}^{-1}$ in the path shown.

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Calculate the magnetic field required to keep the electron in the path.

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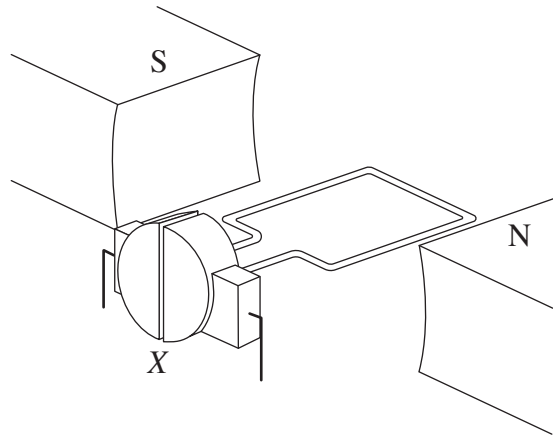
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Question 25 (4 marks)

(a) The diagram represents one type of electric motor.

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Describe the function of part X.

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(b) Explain why the torque of a DC motor decreases as its rotational speed increases.

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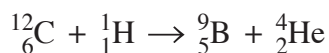
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Question 26 (3 marks)

Consider the following nuclear reaction.

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The masses of the isotopes in this process are shown in the table.

<i>Isotope</i>	<i>Mass (u)</i>
${}^{12}_6\text{C}$	12.064
${}^9_5\text{B}$	9.013
${}^4_2\text{He}$	4.003
${}^1_1\text{H}$	1.008

Calculate the energy released in this reaction.

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Question 27 (8 marks)

- (a) Explain how the composition and temperature of a star can be determined from its spectrum.

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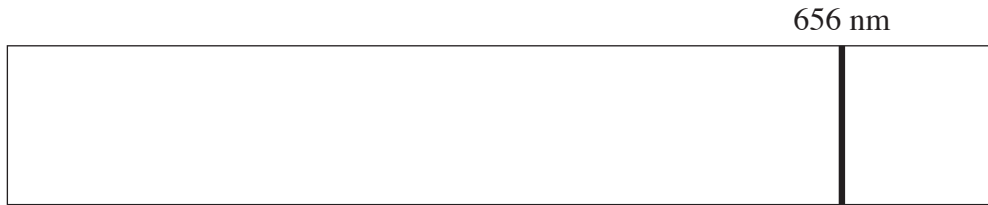
Question 27 continues on page 25

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Question 27 (continued)

- (b) The diagram represents one hydrogen emission line from the spectrum of a star.

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Explain the changes to this spectral line that would be observed as a result of the star's rotational velocity. Modify the diagram to support your answer.

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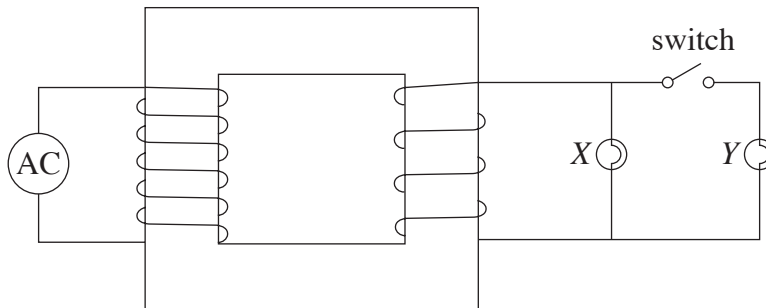
End of Question 27

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Question 28 (5 marks)

An ideal transformer is connected to a 240 V AC supply. It has 300 turns on the primary coil and 50 turns on the secondary coil.

It is connected in the circuit with two identical light globes, *X* and *Y*, as shown.



- (a) Calculate the voltage across light globe *X* when the switch is open.

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- (b) Explain why, after the switch has been closed, the current in the primary coil is different from when the switch is open.

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Question 29 (4 marks)

When light from an incandescent lamp is passed through a plane polarising filter, the intensity of the light is reduced.

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Explain this phenomenon.

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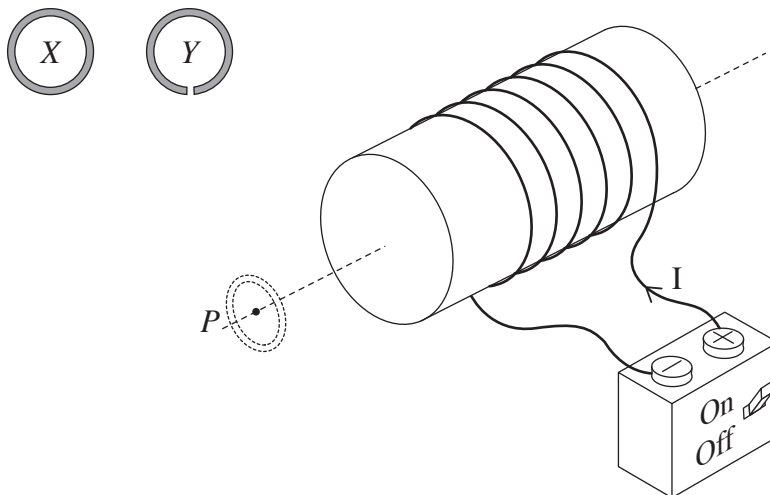
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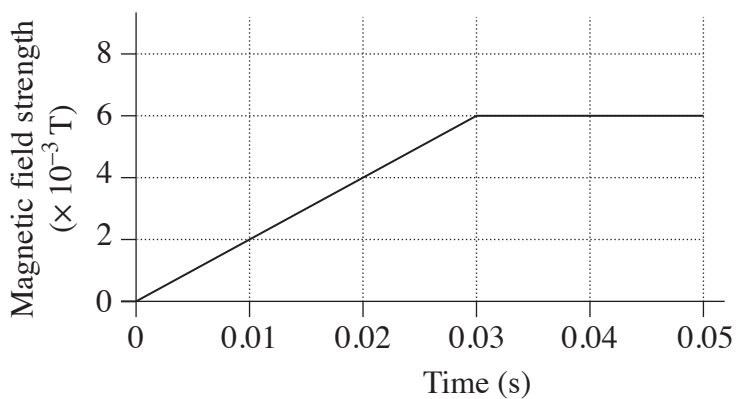
Question 30 (8 marks)

The diagram shows apparatus that is used to investigate the interaction between the magnetic field produced by a coil and two copper rings *X* and *Y*, when each is placed at position *P*, as shown.



Ring *X* is a complete circular ring, and a small gap has been cut in ring *Y*. Each of the rings has a cross-sectional area of $4 \times 10^{-4} \text{ m}^2$.

The power supply connected to the coil produces an increasing current through the coil in the direction shown, when the switch is turned on. This produces a magnetic field at *P* that varies as shown in the graph.



Question 30 continues on page 29

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Question 30 (continued)

- (a) In the first part of the investigation, ring X is held near the end of the electromagnet at position P . 4

Account for the force acting on the ring from 0 to 0.05 seconds after the power supply is turned on.

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- (b) (i) In the second part of the investigation, ring Y is placed at P , and the power supply is turned on. 2

Explain the behaviour of the ring.

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- (ii) Calculate the maximum induced emf in ring Y . 2

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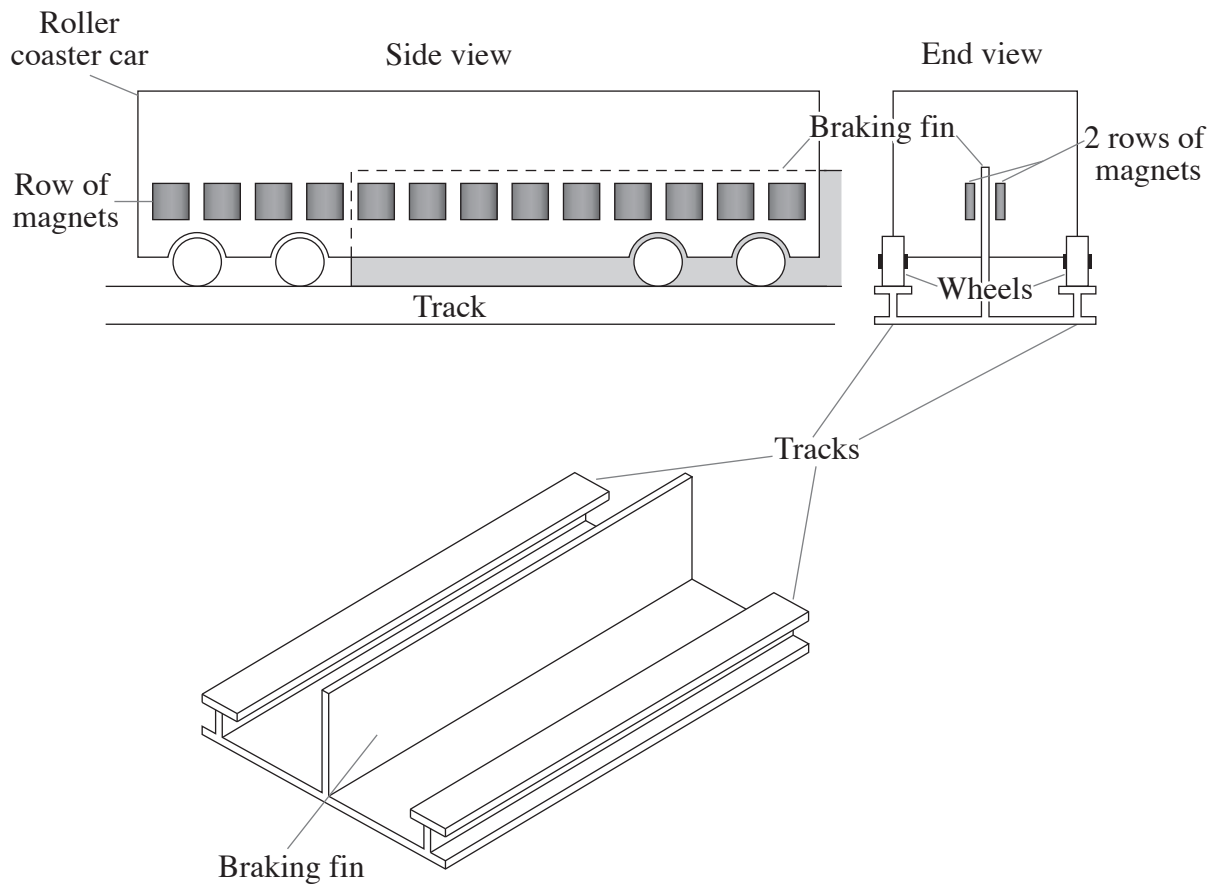
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End of Question 30

Question 31 (5 marks)

A roller coaster uses a braking system represented by the diagrams.

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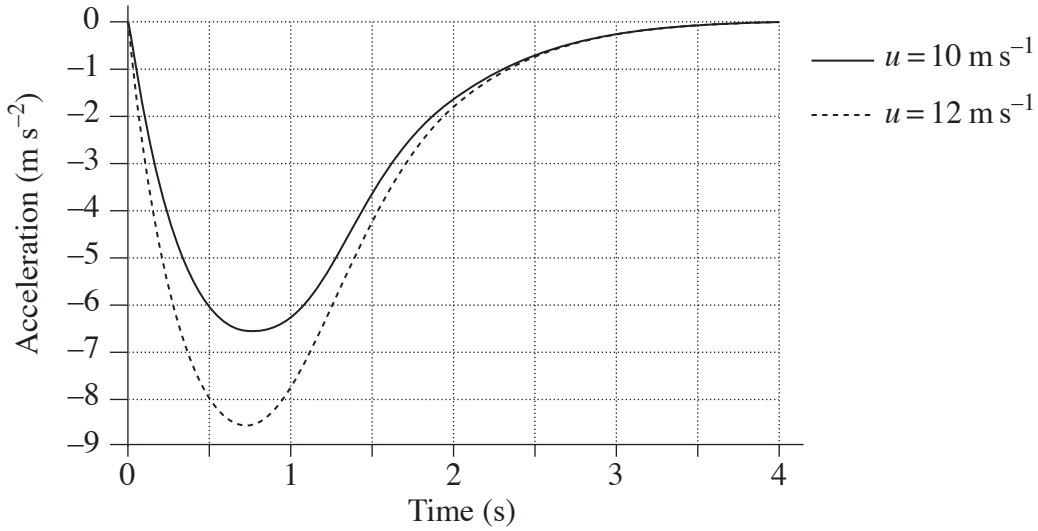


When the roller coaster car reaches the end of the ride, the two rows of permanent magnets on the car pass on either side of a thick aluminium conductor called a braking fin.

Question 31 continues on page 31

Question 31 (continued)

The graph shows the acceleration of the roller coaster reaching the braking fin at two different speeds.



Explain the similarities and differences between these two sets of data.

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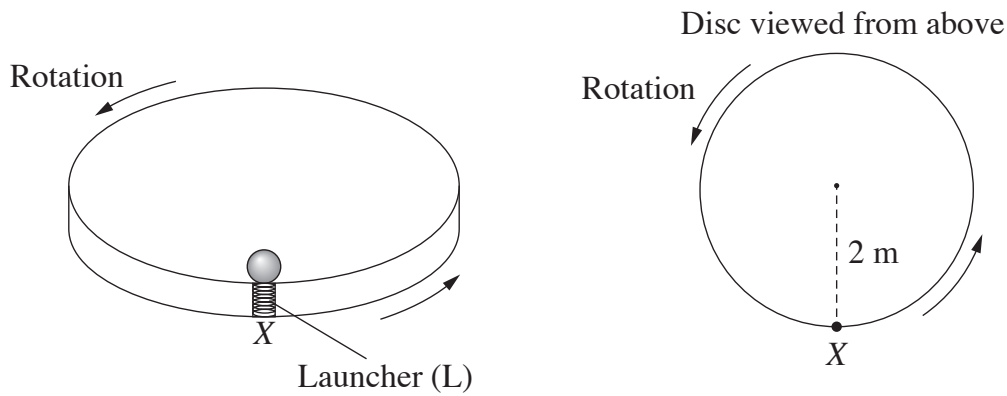
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Question 32 (7 marks)

A horizontal disc rotates at 3 revolutions per second around its centre, with the top of the disc at ground level.

7

At 2 m from the centre of the disc, a ball is held in place at ground level on the top of the disc by a spring-loaded projectile launcher. At position X, the launcher fires the ball vertically upward with a velocity of 5.72 m s^{-1} .



Calculate the ball's position relative to the launcher's new position, at the instant the ball hits the ground.

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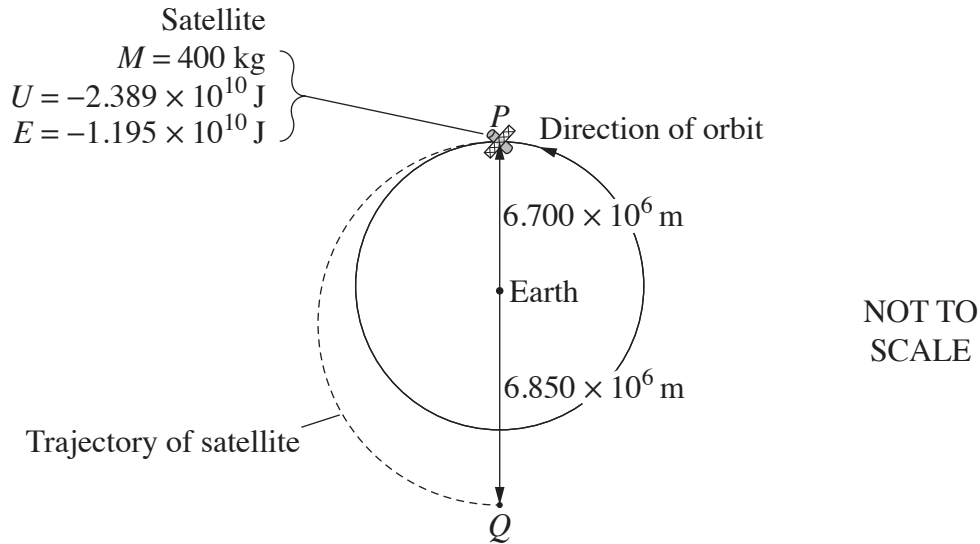
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Question 34 (9 marks)

A 400 kg satellite is travelling in a circular orbit of radius 6.700×10^6 m around Earth. Its potential energy is -2.389×10^{10} J and its total energy is -1.195×10^{10} J.



At point P , the satellite's engines are fired, increasing the satellite's velocity in the direction of travel and causing its kinetic energy to increase by 5.232×10^8 J. Assume that this happens instantaneously and that the engine is then shut down.

The satellite follows the trajectory shown, which passes through Q , 6.850×10^6 m from Earth's centre.

- (a) Analyse qualitatively the energy changes as the satellite moves from P to Q . 2

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Question 34 continues on page 35

Question 34 (continued)

(b) Show that the kinetic energy of the satellite at Q is 1.194×10^{10} J. 4

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(c) Explain the motion of the satellite after it passes through Q . 3

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Section II extra writing space

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Physics

DATA SHEET

Charge on electron, q_e	$-1.602 \times 10^{-19} \text{ C}$
Mass of electron, m_e	$9.109 \times 10^{-31} \text{ kg}$
Mass of neutron, m_n	$1.675 \times 10^{-27} \text{ kg}$
Mass of proton, m_p	$1.673 \times 10^{-27} \text{ kg}$
Speed of sound in air	340 m s^{-1}
Earth's gravitational acceleration, g	9.8 m s^{-2}
Speed of light, c	$3.00 \times 10^8 \text{ m s}^{-1}$
Electric permittivity constant, ϵ_0	$8.854 \times 10^{-12} \text{ A}^2 \text{ s}^4 \text{ kg}^{-1} \text{ m}^{-3}$
Magnetic permeability constant, μ_0	$4\pi \times 10^{-7} \text{ N A}^{-2}$
Universal gravitational constant, G	$6.67 \times 10^{-11} \text{ N m}^2 \text{ kg}^{-2}$
Mass of Earth, M_E	$6.0 \times 10^{24} \text{ kg}$
Radius of Earth, r_E	$6.371 \times 10^6 \text{ m}$
Planck constant, h	$6.626 \times 10^{-34} \text{ J s}$
Rydberg constant, R (hydrogen)	$1.097 \times 10^7 \text{ m}^{-1}$
Atomic mass unit, u	$1.661 \times 10^{-27} \text{ kg}$ $931.5 \text{ MeV}/c^2$
1 eV	$1.602 \times 10^{-19} \text{ J}$
Density of water, ρ	$1.00 \times 10^3 \text{ kg m}^{-3}$
Specific heat capacity of water	$4.18 \times 10^3 \text{ J kg}^{-1} \text{ K}^{-1}$
Wien's displacement constant, b	$2.898 \times 10^{-3} \text{ m K}$

FORMULAE SHEET

Motion, forces and gravity

$$s = ut + \frac{1}{2}at^2$$

$$v^2 = u^2 + 2as$$

$$\Delta U = mg\Delta h$$

$$P = \frac{\Delta E}{\Delta t}$$

$$\sum \frac{1}{2}mv_{\text{before}}^2 = \sum \frac{1}{2}mv_{\text{after}}^2$$

$$\Delta \vec{p} = \vec{F}_{\text{net}} \Delta t$$

$$\omega = \frac{\Delta \theta}{t}$$

$$\tau = r_{\perp} F = rF \sin \theta$$

$$v = \frac{2\pi r}{T}$$

$$U = -\frac{GMm}{r}$$

$$v = u + at$$

$$\vec{F}_{\text{net}} = m\vec{a}$$

$$W = F_{\parallel} s = Fs \cos \theta$$

$$K = \frac{1}{2}mv^2$$

$$P = F_{\parallel} v = Fv \cos \theta$$

$$\sum m\vec{v}_{\text{before}} = \sum m\vec{v}_{\text{after}}$$

$$a_c = \frac{v^2}{r}$$

$$F_c = \frac{mv^2}{r}$$

$$F = \frac{GMm}{r^2}$$

$$\frac{r^3}{T^2} = \frac{GM}{4\pi^2}$$

Waves and thermodynamics

$$v = f\lambda$$

$$f = \frac{1}{T}$$

$$d \sin \theta = m\lambda$$

$$n_x = \frac{c}{v_x}$$

$$I = I_{\text{max}} \cos^2 \theta$$

$$Q = mc\Delta T$$

$$f_{\text{beat}} = |f_2 - f_1|$$

$$f' = f \frac{(v_{\text{wave}} + v_{\text{observer}})}{(v_{\text{wave}} - v_{\text{source}})}$$

$$n_1 \sin \theta_1 = n_2 \sin \theta_2$$

$$\sin \theta_c = \frac{n_2}{n_1}$$

$$I_1 r_1^2 = I_2 r_2^2$$

$$\frac{Q}{t} = \frac{kA\Delta T}{d}$$

FORMULAE SHEET (continued)

Electricity and magnetism

$$E = \frac{V}{d}$$

$$V = \frac{\Delta U}{q}$$

$$W = qV$$

$$W = qEd$$

$$B = \frac{\mu_0 I}{2\pi r}$$

$$B = \frac{\mu_0 NI}{L}$$

$$\Phi = B_{\parallel} A = BA \cos\theta$$

$$\mathcal{E} = -N \frac{\Delta\Phi}{\Delta t}$$

$$\frac{V_p}{V_s} = \frac{N_p}{N_s}$$

$$\vec{F} = q\vec{E}$$

$$F = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2}$$

$$I = \frac{q}{t}$$

$$V = IR$$

$$P = VI$$

$$F = qv_{\perp} B = qvB \sin\theta$$

$$F = I l_{\perp} B = I l B \sin\theta$$

$$\frac{F}{l} = \frac{\mu_0}{2\pi} \frac{I_1 I_2}{r}$$

$$\tau = n l A_{\perp} B = n l A B \sin\theta$$

$$V_p I_p = V_s I_s$$

Quantum, special relativity and nuclear

$$\lambda = \frac{h}{mv}$$

$$K_{\max} = hf - \phi$$

$$\lambda_{\max} = \frac{b}{T}$$

$$E = mc^2$$

$$E = hf$$

$$\frac{1}{\lambda} = R \left(\frac{1}{n_f^2} - \frac{1}{n_i^2} \right)$$

$$t = \frac{t_0}{\sqrt{\left(1 - \frac{v^2}{c^2}\right)}}$$

$$l = l_0 \sqrt{\left(1 - \frac{v^2}{c^2}\right)}$$

$$p_v = \frac{m_0 v}{\sqrt{\left(1 - \frac{v^2}{c^2}\right)}}$$

$$N_t = N_0 e^{-\lambda t}$$

$$\lambda = \frac{\ln 2}{t_{\frac{1}{2}}}$$

PERIODIC TABLE OF THE ELEMENTS

1 H 1.008 Hydrogen		4 Be 9.012 Beryllium		12 Mg 24.31 Magnesium		20 Ca 40.08 Calcium		38 Sr 87.61 Strontium		56 Ba 137.3 Barium		88 Ra Radium		2 He 4.003 Helium	
3 Li 6.941 Lithium		11 Na 22.99 Sodium		19 K 39.10 Potassium		37 Rb 85.47 Rubidium		55 Cs 132.9 Caesium		87 Fr Francium		5 B 10.81 Boron		13 Al 26.98 Aluminium	
7 N 14.01 Nitrogen		15 P 30.97 Phosphorus		31 Ga 69.72 Gallium		49 In 114.8 Indium		81 Tl 204.4 Thallium		113 Nh Nihonium		6 C 12.01 Carbon		14 Si 28.09 Silicon	
9 F 19.00 Fluorine		17 Cl 35.45 Chlorine		33 As 74.92 Arsenic		51 Sb 121.8 Antimony		83 Bi 209.0 Bismuth		115 Mc Moscovium		8 O 16.00 Oxygen		16 S 32.07 Sulfur	
10 Ne 20.18 Neon		18 Ar 39.95 Argon		36 Kr 83.80 Krypton		54 Xe 131.3 Xenon		86 Rn Radon		118 Og Oganesson		7 N 14.01 Nitrogen		15 P 30.97 Phosphorus	
16 S 32.07 Sulfur		34 Se 78.96 Selenium		52 Te 127.6 Tellurium		84 Po Polonium		116 Lv Livermorium		117 Ts Tennessine		6 C 12.01 Carbon		14 Si 28.09 Silicon	
24 Cr 52.00 Chromium		42 Mo 95.96 Molybdenum		74 W 183.9 Tungsten		106 Sg Seaborgium		112 Cn Copernicium		114 Fl Flerovium		29 Cu 63.55 Copper		47 Ag 107.9 Silver	
26 Fe 55.85 Iron		44 Ru 101.1 Ruthenium		76 Os 190.2 Osmium		108 Hs Hassium		110 Ds Darmstadtium		111 Rg Roentgenium		27 Co 58.93 Cobalt		45 Rh 102.9 Rhodium	
28 Ni 58.69 Nickel		46 Pd 106.4 Palladium		78 Pt 195.1 Platinum		110 Ds Darmstadtium		112 Cn Copernicium		114 Fl Flerovium		28 Ni 58.69 Nickel		46 Pd 106.4 Palladium	
30 Zn 65.38 Zinc		48 Cd 112.4 Cadmium		80 Hg 200.6 Mercury		112 Cn Copernicium		114 Fl Flerovium		116 Lv Livermorium		29 Cu 63.55 Copper		47 Ag 107.9 Silver	
32 Ge 72.64 Germanium		50 Sn 118.7 Tin		82 Pb 207.2 Lead		114 Fl Flerovium		116 Lv Livermorium		118 Og Oganesson		30 Zn 65.38 Zinc		48 Cd 112.4 Cadmium	
34 Se 78.96 Selenium		52 Te 127.6 Tellurium		84 Po Polonium		116 Lv Livermorium		118 Og Oganesson		120 Uu Ununnilium		31 Ga 69.72 Gallium		49 In 114.8 Indium	
36 Kr 83.80 Krypton		54 Xe 131.3 Xenon		86 Rn Radon		118 Og Oganesson		120 Uu Ununnilium		122 Uuo Ununnilium		32 Ge 72.64 Germanium		50 Sn 118.7 Tin	
38 Sr 87.61 Strontium		56 Ba 137.3 Barium		88 Ra Radium		110 Ds Darmstadtium		112 Cn Copernicium		114 Fl Flerovium		33 As 74.92 Arsenic		51 Sb 121.8 Antimony	
40 Ca 40.08 Calcium		58 Ce 140.1 Cerium		90 Th 232.0 Thorium		102 No Nobelium		104 Lv Livermorium		106 Lv Livermorium		34 Se 78.96 Selenium		52 Te 127.6 Tellurium	
42 Sr 87.61 Strontium		60 Nd 144.2 Neodymium		92 U 238.0 Uranium		104 Lv Livermorium		106 Lv Livermorium		108 Lv Livermorium		35 Br 79.90 Bromine		53 I 126.9 Iodine	
44 Ru 101.1 Ruthenium		62 Sm 150.4 Samarium		94 Pu 239.0 Plutonium		106 Lv Livermorium		108 Lv Livermorium		110 Lv Livermorium		36 Kr 83.80 Krypton		54 Xe 131.3 Xenon	
46 Pd 106.4 Palladium		64 Gd 157.3 Gadolinium		96 Cm Curium		108 Lv Livermorium		110 Lv Livermorium		112 Lv Livermorium		37 Rb 85.47 Rubidium		55 Cs 132.9 Caesium	
48 Cd 112.4 Cadmium		66 Dy 162.5 Dysprosium		98 Cf Californium		110 Lv Livermorium		112 Lv Livermorium		114 Lv Livermorium		38 Sr 87.61 Strontium		56 Ba 137.3 Barium	
50 Sn 118.7 Tin		68 Er 167.3 Erbium		100 Fm Fermium		112 Lv Livermorium		114 Lv Livermorium		116 Lv Livermorium		39 Y 88.91 Yttrium		57 La 138.9 Lanthanum	
52 Te 127.6 Tellurium		70 Yb 173.1 Ytterbium		102 No Nobelium		114 Lv Livermorium		116 Lv Livermorium		118 Og Oganesson		40 Zr 91.22 Zirconium		58 Ce 140.1 Cerium	
54 Xe 131.3 Xenon		72 Hf 178.5 Hafnium		104 Lv Livermorium		116 Lv Livermorium		118 Og Oganesson		120 Uu Ununnilium		42 Mo 95.96 Molybdenum		60 Nd 144.2 Neodymium	
56 Ba 137.3 Barium		74 W 183.9 Tungsten		106 Lv Livermorium		118 Og Oganesson		120 Uu Ununnilium		122 Uuo Ununnilium		44 Ru 101.1 Ruthenium		62 Sm 150.4 Samarium	
58 Ce 140.1 Cerium		76 Os 190.2 Osmium		108 Hs Hassium		120 Uu Ununnilium		122 Uuo Ununnilium		124 Uuq Ununquadium		46 Pd 106.4 Palladium		64 Gd 157.3 Gadolinium	
60 Nd 144.2 Neodymium		78 Pt 195.1 Platinum		110 Ds Darmstadtium		122 Uuo Ununnilium		124 Uuq Ununquadium		126 Uuq Ununquadium		48 Cd 112.4 Cadmium		66 Dy 162.5 Dysprosium	
62 Sm 150.4 Samarium		80 Hg 200.6 Mercury		112 Cn Copernicium		124 Uuo Ununnilium		126 Uuq Ununquadium		128 Uuq Ununquadium		50 Sn 118.7 Tin		68 Er 167.3 Erbium	
64 Gd 157.3 Gadolinium		82 Pb 207.2 Lead		114 Fl Flerovium		126 Uuq Ununquadium		128 Uuq Ununquadium		130 Uuq Ununquadium		52 Te 127.6 Tellurium		70 Yb 173.1 Ytterbium	
66 Dy 162.5 Dysprosium		84 Po Polonium		116 Lv Livermorium		128 Uuq Ununquadium		130 Uuq Ununquadium		132 Uuq Ununquadium		54 Xe 131.3 Xenon		72 Hf 178.5 Hafnium	
68 Er 167.3 Erbium		86 Rn Radon		118 Og Oganesson		130 Uuq Ununquadium		132 Uuq Ununquadium		134 Uuq Ununquadium		56 Ba 137.3 Barium		74 W 183.9 Tungsten	
70 Yb 173.1 Ytterbium		88 Ra Radium		120 Uu Ununnilium		132 Uuq Ununquadium		134 Uuq Ununquadium		136 Uuq Ununquadium		58 Ce 140.1 Cerium		76 Os 190.2 Osmium	
72 Hf 178.5 Hafnium		90 Th 232.0 Thorium		122 Uuo Ununnilium		134 Uuq Ununquadium		136 Uuq Ununquadium		138 Uuq Ununquadium		60 Nd 144.2 Neodymium		78 Pt 195.1 Platinum	
74 W 183.9 Tungsten		92 U 238.0 Uranium		124 Uuo Ununnilium		136 Uuq Ununquadium		138 Uuq Ununquadium		140 Uuq Ununquadium		62 Sm 150.4 Samarium		80 Hg 200.6 Mercury	
76 Os 190.2 Osmium		94 Pu 239.0 Plutonium		126 Uuo Ununnilium		138 Uuq Ununquadium		140 Uuq Ununquadium		142 Uuo Ununnilium		64 Gd 157.3 Gadolinium		82 Pb 207.2 Lead	
78 Pt 195.1 Platinum		96 Cm Curium		128 Uuo Ununnilium		140 Uuo Ununnilium		142 Uuo Ununnilium		144 Uuo Ununnilium		66 Dy 162.5 Dysprosium		84 Po Polonium	
80 Hg 200.6 Mercury		98 Cf Californium		130 Uuo Ununnilium		142 Uuo Ununnilium		144 Uuo Ununnilium		146 Uuo Ununnilium		68 Er 167.3 Erbium		86 Rn Radon	
82 Pb 207.2 Lead		100 Fm Fermium		132 Uuo Ununnilium		144 Uuo Ununnilium		146 Uuo Ununnilium		148 Uuo Ununnilium		70 Yb 173.1 Ytterbium		88 Ra Radium	
84 Po Polonium		102 No Nobelium		134 Uuo Ununnilium		146 Uuo Ununnilium		148 Uuo Ununnilium		150 Uuo Ununnilium		72 Hf 178.5 Hafnium		90 Th 232.0 Thorium	
86 Rn Radon		104 Lv Livermorium		136 Uuo Ununnilium		148 Uuo Ununnilium		150 Uuo Ununnilium		152 Uuo Ununnilium		74 W 183.9 Tungsten		92 U 238.0 Uranium	
88 Ra Radium		106 Lv Livermorium		138 Uuo Ununnilium		150 Uuo Ununnilium		152 Uuo Ununnilium		154 Uuo Ununnilium		76 Os 190.2 Osmium		94 Pu 239.0 Plutonium	
90 Th 232.0 Thorium		108 Lv Livermorium		140 Uuo Ununnilium		152 Uuo Ununnilium		154 Uuo Ununnilium		156 Uuo Ununnilium		78 Pt 195.1 Platinum		96 Cm Curium	
92 U 238.0 Uranium		110 Lv Livermorium		142 Uuo Ununnilium		154 Uuo Ununnilium		156 Uuo Ununnilium		158 Uuo Ununnilium		80 Hg 200.6 Mercury		98 Cf Californium	
94 Pu 239.0 Plutonium		112 Lv Livermorium		144 Uuo Ununnilium		156 Uuo Ununnilium		158 Uuo Ununnilium		160 Uuo Ununnilium		82 Pb 207.2 Lead		100 Fm Fermium	
96 Cm Curium		114 Lv Livermorium		146 Uuo Ununnilium		158 Uuo Ununnilium		160 Uuo Ununnilium		162 Uuo Ununnilium		84 Po Polonium		102 No Nobelium	
98 Cf Californium		116 Lv Livermorium		148 Uuo Ununnilium		160 Uuo Ununnilium		162 Uuo Ununnilium		164 Uuo Ununnilium		86 Rn Radon		104 Lv Livermorium	
100 Fm Fermium		118 Og Oganesson		150 Uuo Ununnilium		162 Uuo Ununnilium		164 Uuo Ununnilium		166 Uuo Ununnilium		88 Ra Radium		106 Lv Livermorium	
102 No Nobelium		120 Uuo Ununnilium		152 Uuo Ununnilium		164 Uuo Ununnilium		166 Uuo Ununnilium		168 Uuo Ununnilium		90 Th 232.0 Thorium		108 Lv Livermorium	
104 Lv Livermorium		122 Uuo Ununnilium		154 Uuo Ununnilium		166 Uuo Ununnilium		168 Uuo Ununnilium		170 Uuo Ununnilium		92 U 238.0 Uranium		110 Lv Livermorium	
106 Lv Livermorium		124 Uuo Ununnilium		156 Uuo Ununnilium		168 Uuo Ununnilium		170 Uuo Ununnilium		172 Uuo Ununnilium		94 Pu 239.0 Plutonium		112 Lv Livermorium	
108 Lv Livermorium		126 Uuo Ununnilium		158 Uuo Ununnilium		170 Uuo Ununnilium		172 Uuo Ununnilium		174 Uuo Ununnilium		96 Cm Curium		114 Lv Livermorium	
110 Lv Livermorium		128 Uuo Ununnilium		160 Uuo Ununnilium		172 Uuo Ununnilium</									